# 11/26/2022

# brcc keystone logo

Baton Rouge Community College

*Academic Affairs Master Syllabus*

Date Approved: 2 February 2023

Term and Year of Implementation: Spring 2023

**Course Title:** Physical Science II Lab

**BRCC Course Rubric:** PHSC 1031

**Previous Course Rubric**: PHSC 102L

**Lecture Hours per week-Lab Hours per week-Credit Hours**: 0-2-1

**Per semester: Lecture Hours-Lab Hours-Instructional Contact Hours**: 0-30-30

**Louisiana Common Course Number:**

**CIP Code:** 30.0101

**Course Description:** Provides a laboratory component that supplements PHSC 1033 (PHSC 102) content. Covers selected experiments in electricity, magnetism, and chemistry and other physics and physical science phenomena. Not intended for science and engineering majors. This course requires a lab fee.

**Prerequisites:**  None

**Co-requisites:** None

**Suggested Enrollment Cap:** 24

**Learning Outcomes.** *Upon successful completion of this course, the students will be able to:*

1. Use a wide range of experimental apparatus and techniques to make observations and measurements of physics and physical science phenomena in the areas of electricity, magnetism and chemistry.

2. Report experimental data using significant figures and appropriate graphing methods.

3. Use the scientific method to interpret, and statistically evaluate experimental results and to write concise and comprehensive laboratory reports.

**Assessment Measures.** Assessment of all learning outcomes will be measured using the following methods:

1. Administration of unit exams during the semester and a comprehensive final exam at the end of the semester.

2. Instructor-designed assignments including, but not limited to, written and oral assignments, laboratory reports, projects, homework and/or quizzes. All assignments will be graded using an instructor-designed rubric.

**Information to be included on the Instructor’s Course Syllabi:**

* ***Disability Statement*:** Baton Rouge Community College seeks to meet the needs of its students in many ways. See the Office of Disability Services to receive suggestions for disability statements that should be included in each syllabus.
* ***Grading:*** The College grading policy should be included in the course syllabus. Any special practices should also go here. This should include the instructor’s and/or the department’s policy for make-up work. For example in a speech course, “Speeches not given on due date will receive no grade higher than a sixty” or “Make-up work will not be accepted after the last day of class”.
* ***Attendance Policy*:** Include the overall attendance policy of the college. Instructors may want to add additional information in individual syllabi to meet the needs of their courses.
* ***General Policies*:** Instructors’ policy on the use of things such as beepers and cell phones and/or hand held programmable calculators should be covered in this section.
* ***Cheating and Plagiarism*:** This must be included in all syllabi and should include the penalties for incidents in a given class. Students should have a clear idea of what constitutes cheating in a given course.
* ***Safety Concerns:*** In some courses, this may be a major issue. For example, “No student will be allowed in the lab without safety glasses”. General statements such as, “Items that may be harmful to one’s self or others should not be brought to class”.
* ***Library/ Learning Resources:*** Since the development of the total person is part of our mission, assignments in the library and/or the Learning Resources Center should be included to assist students in enhancing skills and in using resources. Students should be encouraged to use the library for reading enjoyment as part of lifelong learning.

**Expanded Course Outline:**

I. Electricity and Magnetism

A. Structure of crystalline solids

B. Ohm’s Law

C. A Study of Series Electric Circuits

D. A Study of Parallel Electric Circuits

E. Alternating Current Impedance

F. Mapping of Magnetic Fields

II. Earth Composition and Structure

III. Heat and Theory of Cooling

IV. Wave Mechanics

A. Optics I

B. Optics II

V. Chemical and Nuclear Reactions

A. Percentage of Oxygen in Potassium Chlorate

B. Avagadro’s Number and the Mole

C. Chemical Formulas and Equations

D. Thickness of a Molecule

E. The Chart of Nuclides